

**Biology
Higher level
Paper 2**

12 May 2026

Zone A morning | Zone B morning | Zone C morning

Session number

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2 hours 30 minutes

Instructions to candidates

- Write your answers in the spaces provided.
- Do not open this paper until you are told to do so.
- Section A: 10 questions. Answer all.
- Section B: 2 questions. Answer both.
- Answers must be written in the spaces provided.
- A calculator is required for this paper.
- The maximum mark for this examination paper is **[80 marks]**.

IBDP 生物

HL

P2 自写答案



Section A

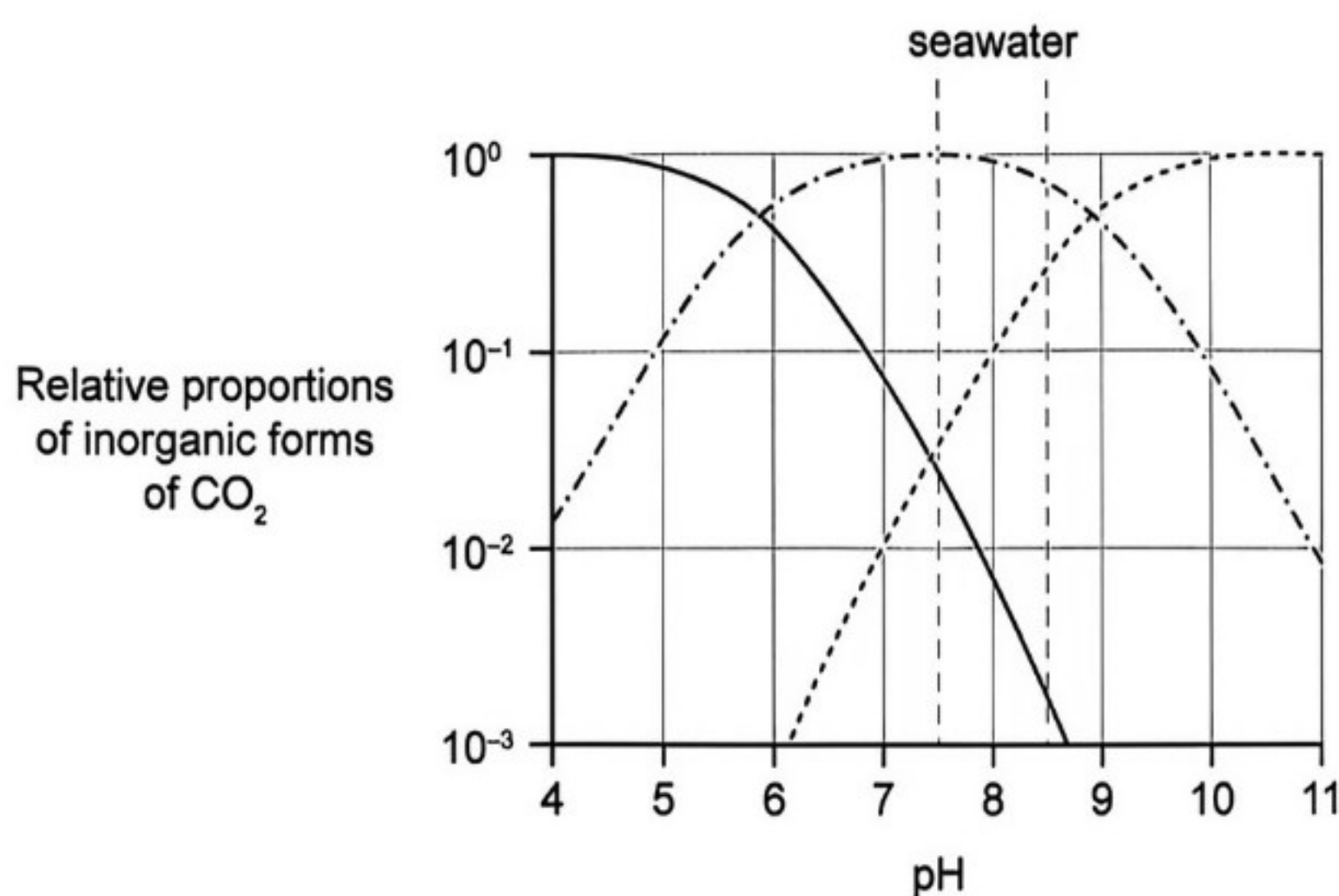
Answer **all** questions. Answers must be written within the answer boxes provided.

1. In the oceans, carbon dioxide (CO_2) dissolved in seawater exists in three main inorganic forms: aqueous CO_2 , bicarbonate ions (HCO_3^-) and carbonate ions (CO_3^{2-}). All three forms of dissolved CO_2 are important for the biological processes of marine organisms. Oceans have an average pH globally of about 8.2. However, pH can vary by ± 0.3 due to local, regional and seasonal factors.

The graph shows the relative proportions of inorganic forms of CO_2 dissolved in seawater at a range of pH levels. The range of pH that is likely to be found in the oceans now and in the future is pH 7.5–8.5.

Key:

— aqueous CO_2 - - - bicarbonate ions (HCO_3^-) . . . carbonate ions (CO_3^{2-})



- (a) Identify the form of dissolved CO_2 that is most abundant in seawater.

[1]

....bicarbonate ions.....

- (b) State the relationship between aqueous CO_2 and carbonate ions (CO_3^{2-}).

[1]

.....inversely related.....
As the relative proportion of aqueous CO_2 increases, the relative proportion of CO_3^{2-} decreases

(This question continues on the following page)



(Question 1 continued)

- (c) Identify **one** natural factor that might influence the pH of seawater. [1]

..... marine organism respiration
 .. (or) Volcanic activity
 (or) photosynthesis of aquatic algae/plants.

Scientists have predicted that the oceans may change in the year 2100 due to CO₂ emissions from industry. The table shows past and two predicted changes to atmospheric CO₂, ocean composition and pH. The predictions were estimated using modelling techniques for the year 2100.

	Pre-industrial (mid-1800s levels)	2005 levels	2x pre- industrial levels	6x pre- industrial levels
Atmospheric concentration of CO ₂ / parts per million	280	380	560	1680
HCO ₃ ⁻ / mol kg ⁻¹	1768	1867	1976	2183
CO ₃ ²⁻ / mol kg ⁻¹	225	185	141	57
Total dissolved inorganic carbon / mol kg ⁻¹	2003	2065	2136	2296
Average pH of oceans	8.18	8.07	7.92	7.49

- (d) Calculate the percentage change of HCO₃⁻ from pre-industrial levels to the 2x pre-industrial levels estimate. [1]

$$\frac{1976 - 1768}{1768} \times 100\% = 11.8\%$$

 11.8%

- (e) Suggest a reason for the researchers modelling 2x and 6x pre-industrial levels estimates for the year 2100. [1]

..... To predict different possible future atmospheric CO₂ emission, assess the severity of ocean acidification.....

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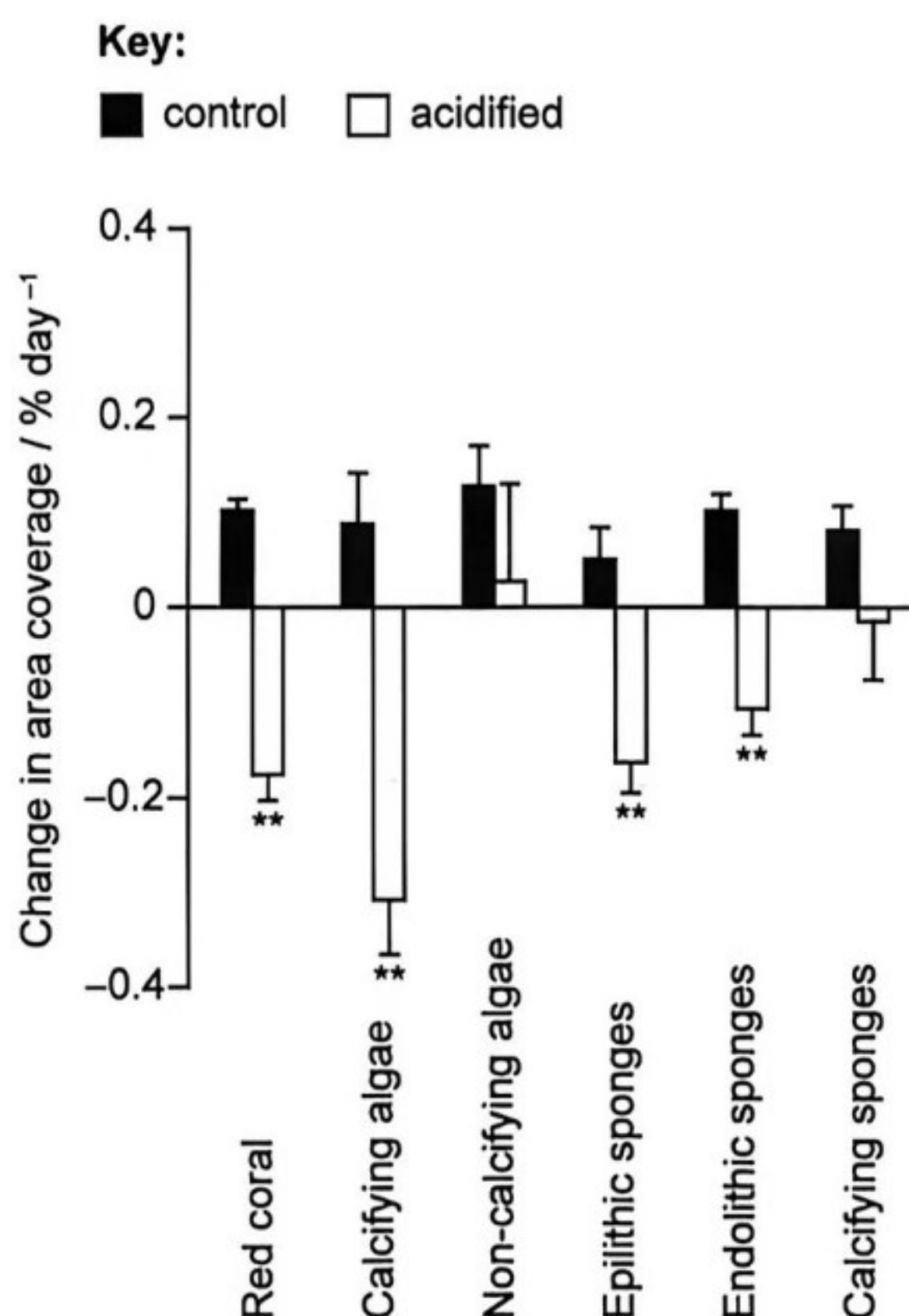


(Question 1 continued)

A separate study investigated the effect of acidification on different reef species in mesocosms.

The graph shows the changes in the growth rates of different reef species when subjected to an acidification treatment and a control treatment. The growth rates are expressed as percentage change in area covered by the species.

Asterisks indicate significant differences (** $p < 0.01$) in the acidified treatment compared with the respective control.



(f) Outline a possible control treatment.

[1]

. Use... seawater... at... normal... ocean... pH, ... all... other... conditions.
 ... are... same... as... the... acidification... treatment.....

(This question continues on the following page)



(Question 1 continued)

- (g) Compare and contrast the effects of the treatments on the calcifying and non-calcifying algae.

[3]

S:
..... Both have lower area coverage under acidified condition.
D:
..... Under acidified condition, Calcifying algae shows negative percentage change, while non-calcifying algae shows positive change.
..... The difference is statistically significant for calcifying algae, but not for non-calcifying algae.

- (h) Using all the data provided, predict the impact of CO₂ levels on reef species by the year 2100.

[3]

..... As CO₂ levels rise and ocean acidification increases, calcifying organisms may experience reduced growth rate due to less CO₃²⁻ ions for shell formation; non-calcifying algae may experience increased growth rate due to more CO₂ for photosynthesis.
..... by 2100, the structure of reef community may change/shift, habitat complexity and biodiversity on coral reefs alter...

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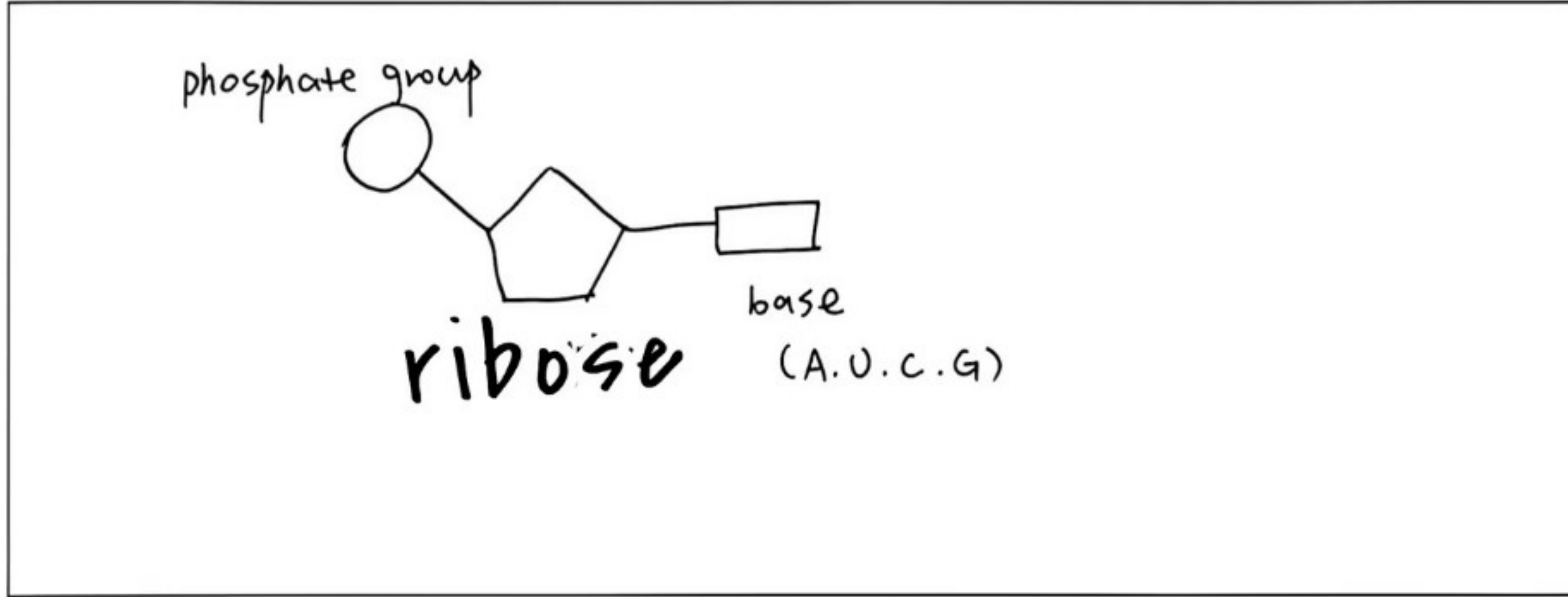
A005



20EP05

Turn over

2. (a) Draw a labelled diagram to show the components of an RNA nucleotide. [3]



- (b) State the type of chemical reaction that joins two nucleotides. [1]

.... condensation reaction

- (c) Outline the production of mRNA by transcription. [3]

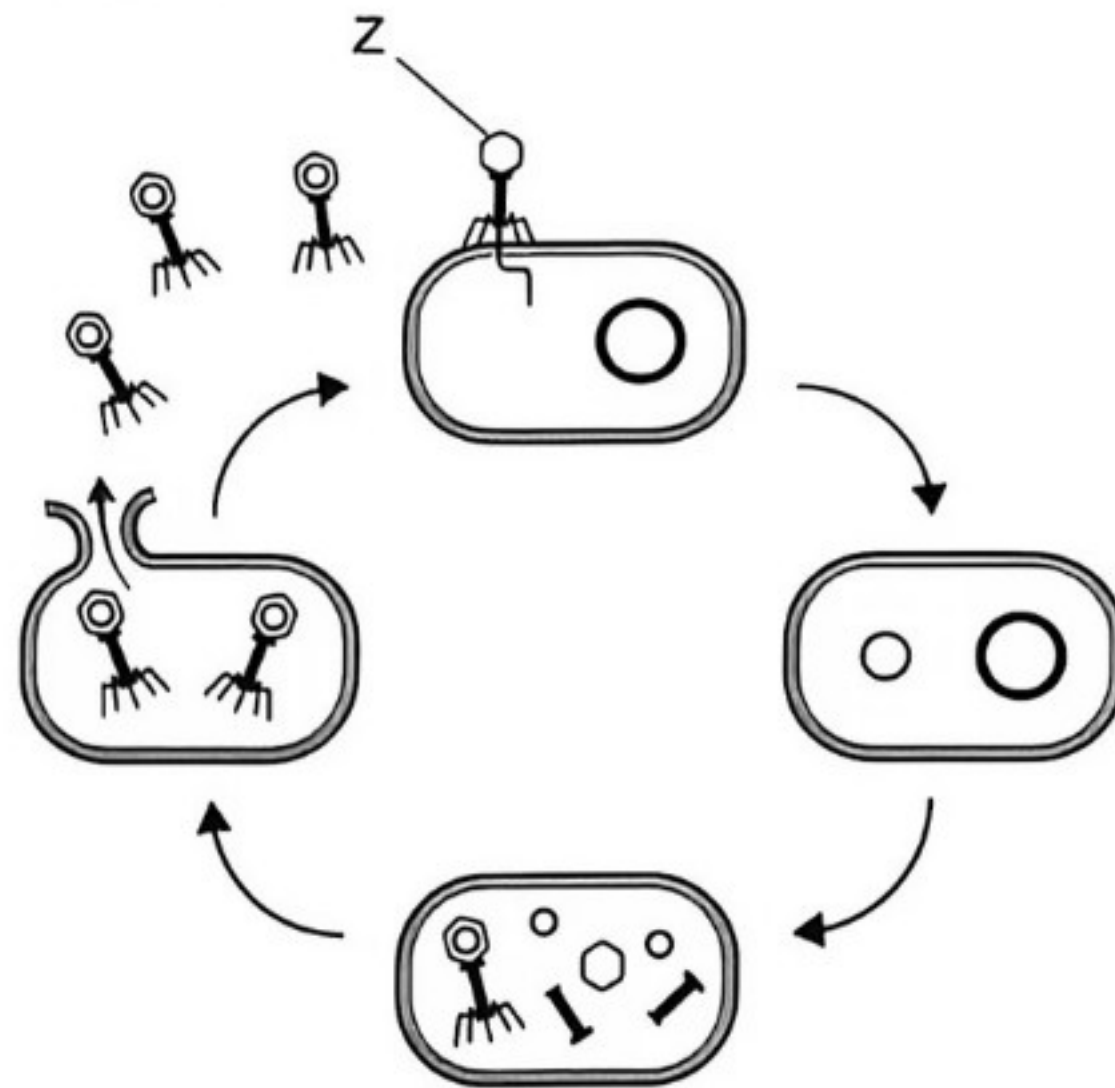
1. Initiation: RNA polymerase binds to promoter of DNA. DNA double strands unwind.....
 2. Elongation: RNA polymerase moves along the template strand in the 3' → 5' direction, synthesising a complementary mRNA strand.....
 3. Termination: transcription ends, the RNA polymerase detaches, DNA rewinds.....

- (d) Describe how mRNA is modified before leaving the nucleus. [2]

1. 5' Capping: A modified guanine nucleotide is added to the 5'-end of the mRNA.....
 2. 3'-end of the mRNA is linked with a poly-A tail.....
 3. RNA splicing: Introns are removed, exons are joined together.



3. The diagram shows viral replication.



(a) State the name of the viral structure labelled Z.

[1]

.... protein ... coat :

(b) Identify the type of life cycle shown in the diagram, giving a reason for your choice.

[2]

.... Lytic ... cycle
 The virus injects its genetic material, assembles new viral particles
 and causes lysis of the host cell
 No evidence of integration into the host genome

(c) State the mode of existence of all viruses.

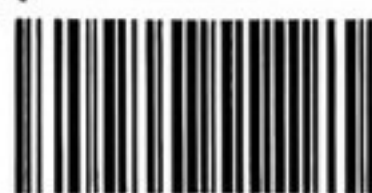
[1]

.... ~~The~~ ... Obligate ... intracellular ... parasites,

(d) Discuss the evidence for several origins of viruses.

[2]

... VIRUS - first hypothesis : Viruses in evolved before or co-evolved with
 their current host cells
 ... Escaped gene hypothesis : viruses arose from genetic elements that
 gained the ability to move
 Regressive / Reduction hypothesis : viruses are remnants of cellular
 organisms .



4. Compare and contrast the organelles in animal and fungal cells.

[4]

	Animal cells	fungal cells
S:	both have cell wall cytoplasm, cell membrane, nucleus,	
D:	both have ribosomes, mitochondria,	
	both have ER and Golgi apparatus,	
	have no cell wall,	have cell wall
	have lysosomes	have no lysosomes
	have centrioles	have no centrioles

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A005



20EP08



5. (a) The table shows some information about the endocrine system. State the gland, hormone and function missing from the table. [3]

Gland	Hormone	Function
pineal	 melatonin	circadian rhythm
. Adrenal . gland	epinephrine (adrenaline)	fight-or-flight response
pituitary	antidiuretic hormone	Reduce urine production .

- (b) The pancreas produces the hormone glucagon. Explain how glucagon is used to maintain constant blood glucose levels. [3]

.... When blood glucose concentration drops, 2 cells in the pancreas detect the change and secrete glucagon into the bloodstream .
 Glucagon acts on liver cells, stimulating the breakdown of glycogen into glucose, and promoting gluconeogenesis, therefore, blood glucose levels raise back to normal

1
2
3



6. The CCR5 Δ 32 allele has been linked to resistance to bubonic plague, although these claims are controversial.

(a) State the cause of a new allele forming. [1]

... Mutation

(b) The infectious disease bubonic plague transfers from other species to humans. State the name for this type of disease. [1]

.... Zoonose

(c) The frequency of the CCR5 Δ 32 allele is higher in areas where bubonic plague is common. Explain the reason for this. [3]

..... In areas where bubonic plague is common, individuals carrying the CCR5 Δ 32 allele have a survival advantage because they are more resistant to the disease. So they are more likely to survive, reproduce and pass on their allele to their offspring. Over generations, the frequency of this allele increases. This process is natural selection.



7. (a) State the name for a genotype of individuals with one dominant and one recessive allele. [1]

.....heterozygous.....

- (b) Distinguish between dominant and recessive alleles. [1]

represented by letter in upper case lower case
has the same effect on the has the same effect on the phenotype
phenotype whether it's present only when present in the homozygous state
in homozygous or heterozygous state

- (c) Carriers are individuals who possess one allele for a genetic condition but have no symptoms of that condition. Phenylketonuria (PKU) is a human disease caused by a recessive allele (d). Construct a Punnett grid to show the probability that two carrier parents will have a child with PKU. [2]

Parental generation

carrier
Dd

carrier
Dd

Gametes

(D) (d)

(D) (d)

	$\frac{1}{2}D$	$\frac{1}{2}d$
$\frac{1}{2}D$	$\frac{1}{4}DD$ normal	$\frac{1}{4}Dd$ normal
$\frac{1}{2}d$	$\frac{1}{4}Dd$ Normal	$\frac{1}{4}dd$ PKU

offspring

genotypes

DD

Dd

dd

1

:

2

:

1

phenotypes

Normal

:

PKU

3

:

1

- (d) Haemophilia is a genetic disorder. Explain the reason that males cannot be carriers of this condition. [2]

.....Haemophilia is caused by a recessive allele in X chromosome
.....Males have sex chromosomes as X and Y.....
.....If a male has a dominant allele in X, he is not affected..
.....If a male has a recessive allele in X, he is affected.....



Section B

Answer **two** questions. One additional mark is available for the construction of your answers for each question. Answers must be written within the answer boxes provided.

8. Blood transports gases such as carbon dioxide and oxygen.
- (a) Outline the structure of a haemoglobin molecule. [3]
 - (b) Explain how the Bohr shift allows more oxygen to be supplied to actively respiring tissues. [8]
 - (c) Describe the adaptations of red blood cells (erythrocytes) to their function. [4]
9. Animals are adapted to survive in many habitats.
- (a) Outline how animal cell membranes are adapted to habitats of different temperatures. [3]
 - (b) Describe the adaptations of herbivores for feeding on plants and of plants for resisting herbivory. [5]
 - (c) Explain how the abiotic features of a hot desert biome result in **three** adaptations of a **named** animal for survival. [7]
10. Membranes are important for allowing substances to move in and out of cells and organelles.
- (a) Outline the changes that occur to plant cells when plant tissue is bathed in a hypotonic solution. [4]
 - (b) Describe how a resting potential is established and maintained. [4]
 - (c) Explain the role of the electron transport chain in the generation of ATP by cellular respiration. [7]

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A005



20EP12



8. a. - protein / conjugated protein
- - four peptide chains / two α -chains and two β -chains
- - each chain is held around a haem group
- - each haem group has an atom of iron
- b. - In respiring tissues, the concentration is high
- - oxygen is released from the oxyhaemoglobin more
- readily
- - due to protons from the dissociation of carbonic
- acid
- - So the increase in carbon dioxide concentration shifts
- the oxygen dissociation curve to the right
- - This benefits respiring cells with increased metabolism
- - Greater amounts of CO_2 are released
- - Hb releases its O_2 at regions of greatest
- respiratory rate
- - O_2 is needed for aerobic cell respiration
- c. - small size, about $8\text{ }\mu\text{m}$, fit through the small
- lumen of capillaries
- - ~~small~~ small size, have a large SA:V ratio,
- allowing fast diffusion
- - biconcave shape, short diffusion distance
- - lack of nucleus, more room for haemoglobin

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9.a. In cold environments

- - higher proportion of unsaturated fatty acids
- - The kinks prevent tight packing
- - contains more cholesterol
- - maintain membrane fluidity at low temperature

In warm environments

- - higher proportion of saturated fatty acids
- - tightly pack
- - prevent the membrane from being too fluid at high temperature

..... for
b. Adaptations of herbivores feeding on plants

- - teeth: Incisors for cutting plant materials
- Large flat molars for grinding
- - Digestive system: long tracts
- four chambered stomach (Ruminants)
- mutualistic bacteria for digesting cellulose

..... Adaptations of plants for resisting herbivory

- - mechanical defense: spines, thorns, tough leaves
- - chemical defense: toxins, bitter compounds
- - indirect defense: matter to attract predators of herbivores

c. Abiotic features of a ~~hot~~ hot desert biome

- - high temperatures, strong solar radiation
- - low water availability
- - large temperature fluctuations

..... Adaptations of camels

- - can tolerate a wide of body temperatures
- - sweat little
- - can drink large amount of water
- - store fat in humps, produce water from metabolizing fats
- - produce highly concentrated urine and dry feces



10. a. A hypotonic solution has a higher water potential than the cell contents.

Water moves into the cell by osmosis down the water potential gradient.

The vacuole swells, cell membrane/cytoplasm pushes against the cell wall.

The plant cell becomes turgid.

The cell wall prevents the cell from bursting.

No net water movement once the turgor pressure

balances the osmotic influx.

b. Sodium-potassium pumps actively transport 3 Na^+ out of the cell and 2 K^+ into the cell.

ATP is used.

This creates a concentration gradient.

The membrane is more permeable to K^+ than Na^+ at rest.

K^+ diffuse out of the cell.

The outward movement of K^+ makes the inside of the cell negatively charged relative to the outside.

c. ETC is located in the inner membrane of mitochondria.

Glycolysis and Krebs cycle produce NADH and FADH_2 .

NADH and FADH_2 release electrons.

As electrons pass through protein complexes, energy is released.

The released energy is used to pump protons from the

matrix into the intermembrane space.

An electrochemical gradient is created.

Chemiosmosis

Protons diffuse back into the matrix through ATP synthase.

The flow of protons provides energy for ATP synthesis.

Oxygen acts as the final electron acceptor, to

form water.

Which prevents the ETC from being blocked.

